



# Interfacially reinforced carbon fiber/epoxy composite laminates via in-situ synthesized graphitic carbon nitride (g-C<sub>3</sub>N<sub>4</sub>)

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## ABSTRACT

Carbon fiber composite laminates were interfacially reinforced through in-situ synthesis of g-C<sub>3</sub>N<sub>4</sub> on the carbon fibers. The introduced g-C<sub>3</sub>N<sub>4</sub> greatly improved the roughness, functional groups and wettability on the carbon fiber surface and markedly enhanced the interfacial properties of composite laminates. The surface free energy of carbon fibers was increased by 67.81%. Interlaminar shear strength and interfacial shear strength of composite laminates were increased from 51.84 to 72.09 MPa and 44.62–73.41 MPa, respectively. The significantly enhanced interfacial properties enabled the mechanical performance of composite laminates to reach a superior state. Tensile strength and bending strength were increased by 19.54 and 10.51%, respectively. The total absorbed energy of impact experiment was also enhanced from 1.14 to 1.78 J. Meanwhile, dynamic mechanical properties and hydrothermal aging resistance were also ameliorated significantly. The improved interfacial properties and mechanical properties were ascribed to the increased mechanical interlocking, enhanced chemical bonding and ameliorated wettability created by g-C<sub>3</sub>N<sub>4</sub>.

## 1. Introduction

Carbon fiber reinforced epoxy composites (CFRECs) as ideal structural materials with the merit of high strength-to-weight ratio and rigidity have now been widely used in aerospace, civil engineering, automotive, sports goods and other field [1–3]. In general, the ultimate performance of composites is largely dependent on reinforcement, matrix and interphase [4–7]. Interphase as a “bridge” of bonding and load transfer between carbon fiber (CF) and epoxy resin has a crucial influence on the ultimate performance of CFRECs [8–12]. However, the

poor interfacial compatibility between CF and epoxy resin results in low interfacial strength and impact toughness, which severely limit further development and applications of CFRECs [13–18]. For the past thirty years, considerable research efforts have been made to improve the interfacial compatibility between CF and epoxy resin. Constructing interphase containing nanomaterials is considered as an effective method to solve this problem [19–22]. Various strategy have been employed to create interphase containing nanomaterials, such as coating with sizing agent [23–27], in-situ synthesis [28–30], chemical grafting [31–33] and electrochemical deposition (ECD) [34–37]. Among those

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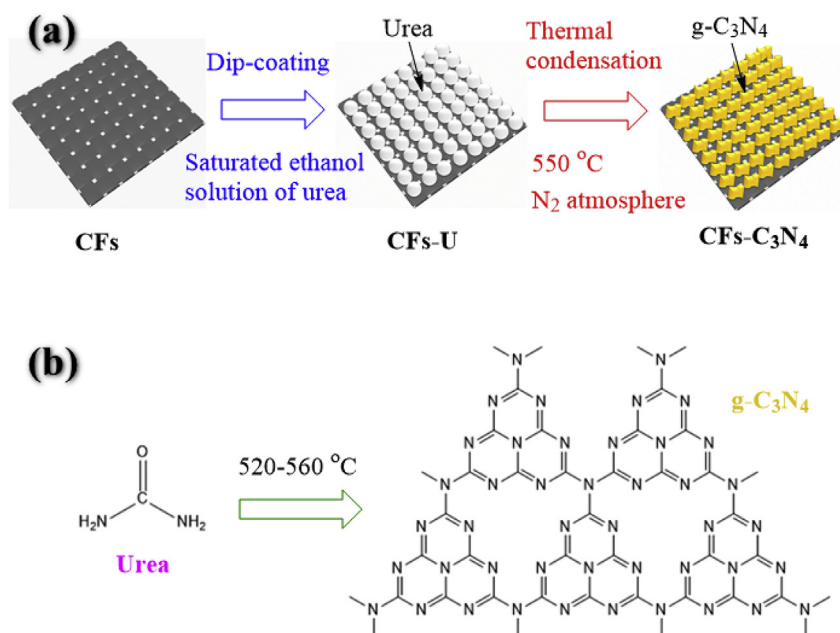


Fig. 1. Schematic illustration of (a) in-situ synthesis of g-C<sub>3</sub>N<sub>4</sub> on CFs surface; (b) synthesis process of g-C<sub>3</sub>N<sub>4</sub> by thermal polymerization of urea.

approaches, in-situ synthesis can regulate the interphase structure more effectively, ameliorate the interfacial compatibility and strengthen the interfacial adhesion.

The properties of the interphase containing nanomaterials are dependent to a great extent on the inherent properties of the nanomaterials [38–40]. In previous research reports, carbon nanotubes (CNTs) [41–43], graphene/graphene oxide (GO) [44–47], ZnO nanowires [48,49] and TiO<sub>2</sub> nanorods [50,51] etc. were successfully applied in the construction of interphase of composites. For example, Sharma and his co-worker [52] synthesized CNTs and nanofibers on CF substrate using thermal chemical vapor deposition (CVD) process. The composites made of CNTs coated CF showed 69% higher tensile strength as compared to the composites made of original CFs. Wang et al. [53] found that the grafting force between CNTs and CF created by CVD was so strong (up to 5 μN) that the interfacial shear strength of CF composites was increased by 30%. Zheng et al. [54] used ZnO nanowires to construct the interphase of CF composites through a facial hydrothermal method. The interfacial shear strength of CF composites was increased by 64% after the in-situ growth of ZnO nanowires on the CF surface. Due to the rigid ZnO nanowires that can more effectively improve the toughness of CF composites compared to those flexible fillers (like CNTs) with easy collapse during epoxy infusion, both mode I and mode II interlaminar toughness of CF composites laminates were increased by 63% and 22%, respectively. Meanwhile, the introduction of nanomaterials can significantly change the physical and chemical properties of interphase. Harpreet et al. [8] confirmed that the relative interphase size and interfacial area were increased by approximately 3 times in the presence of CNTs. The gradual variation in modulus from epoxy to CF through the interphase region containing CNTs effectively suppressed the unwanted stress concentrations around the CF due to the existence of sharp interface, thus markedly enhanced the stiffness, IFSS and load bearing capacity of CF composites. Although the research on interfacially reinforced CF composites by constructing interphase containing nanomaterials has made many significant progress, enormous potential of interphase containing nanomaterials has not been fully exploited. Some new materials should be explored in the construction of interphase of CF composites.

In this study, graphitic carbon nitride (g-C<sub>3</sub>N<sub>4</sub>) as a conjugated polymer with graphite-like structure was used to construct the interphase of CF composites via an in-situ synthesis process. The interphase

created by g-C<sub>3</sub>N<sub>4</sub> markedly improved the compatibility between CF and matrix, and endowed CF composites with superior interfacial properties than that of other CF composites. Meanwhile, the surface morphology and fracture appearance of CF and its composites were observed by scanning electron microscope (SEM). Surface chemical elements, functional group and phase structure of CF and g-C<sub>3</sub>N<sub>4</sub> were investigated by X-ray photoelectron spectroscopy (XPS), Fourier Transform Infrared Spectroscopy (FTIR) and X-ray diffraction (XRD), respectively. Wettability studies were carried out on a dynamic contact angle analysis (DCAT). The interfacial compatibility between CF and epoxy was evaluated by interfacial shear strength (IFSS), interlaminar shear strength (ILSS), impact toughness and hydrothermal aging resistance. Moreover, the practical application of CF composites was also explored by dynamic thermomechanical analysis (DMA).

## 2. Materials and experiments

### 2.1. Materials

The plain weave carbon fiber fabric (T300 3K, diameter: 7 μm, surface density: 200 g/cm<sup>2</sup>) used in this study was purchased from Guangwei Co. Ltd (China). The carbon fiber fabric was thermally treated at 450 °C for 1 h in N<sub>2</sub> environment to remove the sizing (the obtained samples were denoted as CFs). Epoxy resin (HT-723A-1) and curing agent (HT-723B-1) were supplied from Wells Advanced Materials (shanghai) Co., Ltd (China). Urea, absolute ethanol and acetone were purchased from Aladdin (China). All chemicals, reagents and solvents were of analytical grade and used as received without further purification.

### 2.2. Surface treatment of CFs fabric

The in-situ synthesis of g-C<sub>3</sub>N<sub>4</sub> on carbon fiber fabric was realized through a combination of dip-coating and thermal condensation method. The schematic formation of g-C<sub>3</sub>N<sub>4</sub> on CFs surface is shown in Fig. 1(a). A typical synthesis process is as follows. First, the CFs were immersed in saturated ethanol solution of urea for 40 min and dried at 80 °C for 5 h (the CFs coated with urea were called as CFs-U). Then the CFs-U were embedded into powders of urea and calcined at 550 °C for 4 h under a N<sub>2</sub> atmosphere (the synthesis process of g-C<sub>3</sub>N<sub>4</sub> was

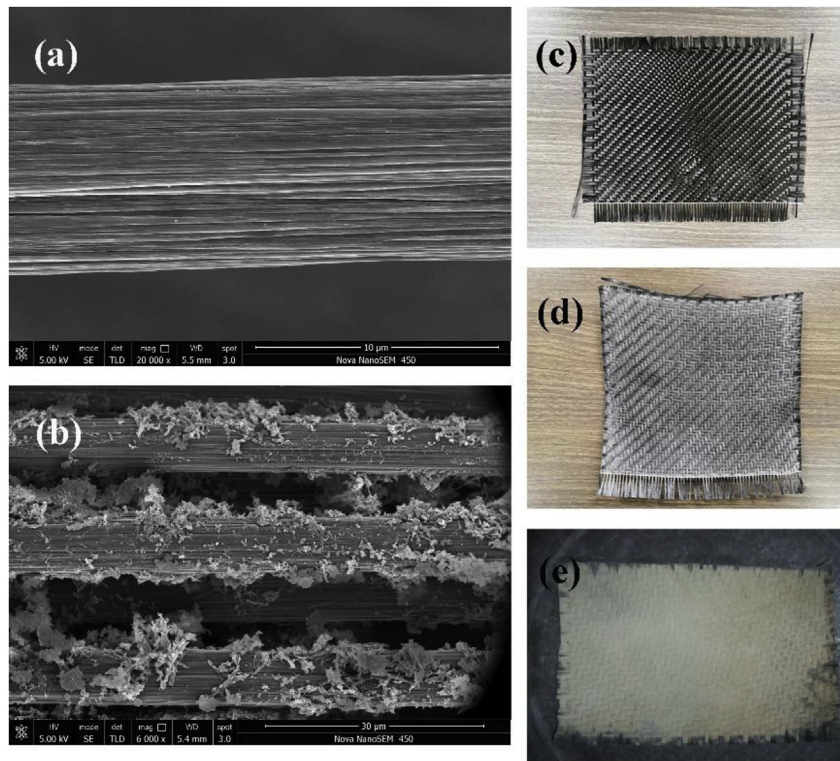


Fig. 2. Surface morphology of carbon fiber fabric: (a)/(c) CFs; (b)/(d) CFs-C<sub>3</sub>N<sub>4</sub>; (e) CFs-U.

illustrated in Fig. 1(b)). This as-prepared CFs fabric was ultrasonically washed in absolute ethanol to remove the loosened g-C<sub>3</sub>N<sub>4</sub> on the CF surface, followed by drying at 80 °C for 4 h in a vacuum drying oven (the obtained samples were named as CFs-C<sub>3</sub>N<sub>4</sub>). Moreover, pure g-C<sub>3</sub>N<sub>4</sub> was also synthesized under the same conditions.

### 2.3. Preparation of CF composite laminates

CF composite laminates comprising of eight plies of plain weave CF fabric (with/without g-C<sub>3</sub>N<sub>4</sub>) were fabricated via a vacuum bag-pressure moulding. The CF composite laminates were pre-formed by vacuum bag moulding and left to cure at 50 °C for 2 h. Then the laminates were post-cured via pressure moulding at 90 °C for 3 h (moulding pressure for 10 MPa). The cured laminates were respectively machined into ASTM D3039 tensile coupons (175 mm length × 25 mm width × 2 mm thick), ASTM D7264 flexural coupons (155 mm length × 13 mm width × 4 mm thick) and ASTM D2344 shear coupons (24 mm length × 8 mm width × 4 mm thick) using a diamond saw.

### 2.4. Characterizations

The surface morphologies of CF fabric and fracture surface features were detected by SEM (HITACHI, FE-SEM SU8000). The chemical binding states of CF fabric were characterized by X-ray photoelectron spectroscopy (XPS, K-Alpha Compact, Thermo Scientific). The surface functional groups of CF fabric were investigated by Fourier transform infrared spectroscopy (FTIR, VERTEX70, Bruker). The phase structures of CF and g-C<sub>3</sub>N<sub>4</sub> were characterized using X-ray diffraction (XRD, Rigaku D/max-rB). Thermogravimetry (TG) analysis of CFs was carried out on a DSC/DTA TG simultaneous thermal analyzer (STA 449 F5 Jupiter, NETZSCH) from 30 to 800 °C at a heating rate of 10 °C/min.

The wettability between CF and epoxy was evaluated by dynamic contact angle test using a dynamic contact angle meter and tensiometer (DCAT21, DataPhysics Instruments). Deionized water ( $\gamma_s^d = 21.8$  mN/m,  $\gamma_s = 72.8$  mN/m) and diiodomethane ( $\gamma_s^d = 50.8$  mN/m,  $\gamma_s$

= 50.8 mN/m) were used as test liquids for the determination of surface energies ( $\gamma_s$ ), dispersive component ( $\gamma_s^d$ ) and polar component ( $\gamma_s^p$ ) of CFs.

The mechanical properties were evaluated using a universal testing machine (E45.305, MTS). The tensile strength of CF monofilament was tested in accordance with ASTM D3379-75 (no fewer than 60 samples were evaluated), and the obtained data were processed using Weibull statistical method. The interfacial shear strength (IFSS) of CF composites was measured by micro-droplet experiment, which was performed on an evaluation device for interfacial adhesion of composites (HM410, MODEL) at the crosshead speed of 0.08 mm/s. The IFSS was calculated by Eq. (1), where  $d$ ,  $L$  and  $F$  represent diameter of CF monofilament, embedded length of resin droplet and maximum pull-out force, respectively.

$$IFSS = \frac{F}{\pi d L} \quad (1)$$

The interlaminar shear strength (ILSS) of CF composite laminates was evaluated by a three-point short beam bending according to ASTM D2344 standard at a crosshead speed of 2 mm/min. Then ILSS was calculated according to Eq. (2), where  $P_b$ ,  $h$  and  $d$  represent the breaking load, width and thickness of specimen, respectively. The tensile test was implemented on the basis of ASTM D3039 at a constant crosshead rate of 2 mm/min. The flexural properties were obtained by three-point bending according to ASTM D7264 standard at a constant rate of crosshead movement of 1.0 mm/min. Each used data for mechanical properties of CF composite laminates was an averaged value of 6 specimens.

$$ILSS = \frac{3P_b}{4bh} \quad (2)$$

The impact toughness was assessed using a pendulum impact test system (9250HV, Instron). The span of tested samples was set to 40 mm, the weight and velocity of pendulum were selected as 3 kg and 2 m/s, respectively. Each specimen was repeatedly tested at least six times. Dynamic mechanical experiment was implemented on a dynamic

mechanical thermal analyzer (Q800, TA) operating in the three-point bending mode at a frequency of 1.0 Hz. The dimensions of specimen was 60 mm × 10 mm × 2 mm. The temperature range of experiment was limited to 30–180 °C with a heating rate of 3 °C/min. The hydrothermal aging resistance of CF composite laminates was studied by tracing the change of ILSS after laminates immersed in the boiling water for 48 h. At least 5 samples were tested for the evaluation of hydrothermal aging resistance.

### 3. Results and discussion

#### 3.1. Surface morphology and crystal phase of CF fabric

The surface morphology of carbon fiber fabric was illustrated in Fig. 2. The surface of untreated CFs was smooth and neat, and some grooves along the direction of fiber axis were uniformly distributed throughout the carbon fiber surface. After surface treatment, some marked changes of surface morphology can be found on the CFs-C<sub>3</sub>N<sub>4</sub> surface. Some micron-sized g-C<sub>3</sub>N<sub>4</sub> particles were uniformly distributed on the surface of CFs-C<sub>3</sub>N<sub>4</sub>. Moreover, the appearance characteristics of carbon fiber fabric was also obtained by electronic photos (as shown in Fig. 2 (c)–(d)). Originally, the surface of CFs was neat and bright. The surface of CFs-U was then coated by a homogeneous layer of white urea after impregnation treatment. Lastly, the urea layer on carbon fiber fabric was changed into light-yellow g-C<sub>3</sub>N<sub>4</sub> through a thermal polymerization. The existence of g-C<sub>3</sub>N<sub>4</sub> particles greatly increased the surface roughness of carbon fibers and interfacial contact area in carbon fiber composite laminates.

Besides, the structure and composition of carbon fiber fabric were

confirmed by XRD. As shown in Fig. 3(a), the observed only broad characteristic diffraction peak at around 23.0° is assigned to the (002) plane of graphitic structure of CFs [55]. Pure g-C<sub>3</sub>N<sub>4</sub> has two characteristic diffraction peaks at 13.1° and 27.4°, respectively. The diffraction peak at 13.1° is attributed to the (100) plane resulting from the in-plane ordering of tri-s-triazine units with a distance of 0.670 nm, and the diffraction peak center at 27.4° is normally associated with the (002) plane due to the interlayer stacking reflection of conjugated aromatic structures with an interlayer space of 0.323 nm [56]. In comparison with CFs and pure g-C<sub>3</sub>N<sub>4</sub>, CFs-C<sub>3</sub>N<sub>4</sub> exhibits the same diffraction peaks with pure g-C<sub>3</sub>N<sub>4</sub> due to a large amount of g-C<sub>3</sub>N<sub>4</sub> existing on the CFs-C<sub>3</sub>N<sub>4</sub> surface.

#### 3.2. Surface elemental composition of carbon fiber fabric

FTIR analysis in Fig. 3(b) reveals the surface functional groups of carbon fiber fabric and g-C<sub>3</sub>N<sub>4</sub>. No any significant characteristic absorption peak can be found on the FTIR curve of CFs, which means that the CF surface is inert. After impregnation treatment, the characteristic absorption peaks of urea appeared on the FTIR spectra of CFs-U. The characteristic absorption peaks centered at 1160 and 1450 cm<sup>-1</sup> corresponded to the vibration mode and stretching mode of C–N bond, respectively [57]. The absorption peaks located in 1680–1630 cm<sup>-1</sup> were attributed to the stretching vibration of C=O bond. The characteristic absorption peaks in the range of 900–650 and 3600–3100 cm<sup>-1</sup> reflect the in-plane bending vibration and stretching vibration of N–H bond, respectively [57]. For the FTIR spectra of pure g-C<sub>3</sub>N<sub>4</sub>, the characteristic absorption peaks at around 1700–1200 cm<sup>-1</sup> represent the stretching vibration of conjugated CN rings, and the

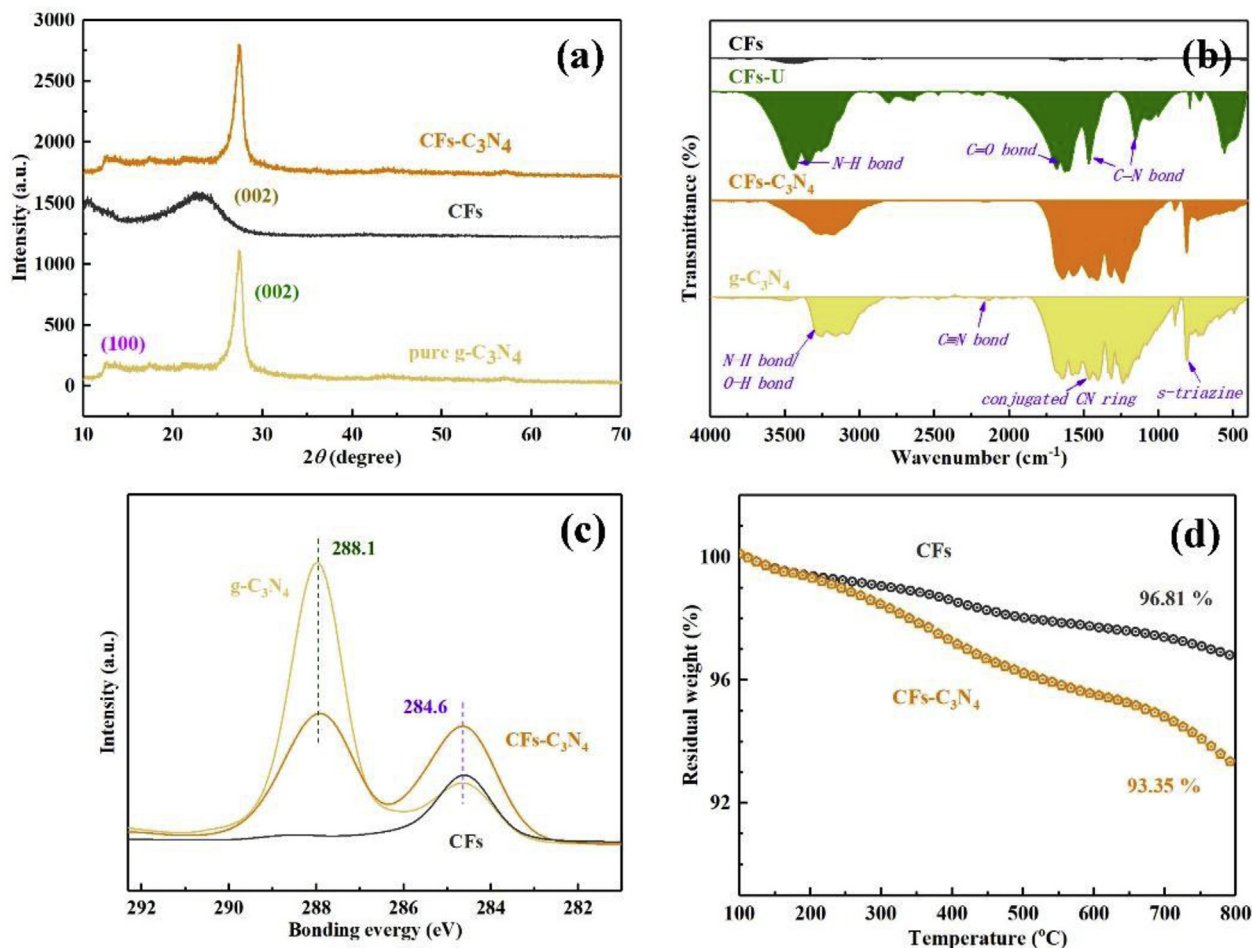


Fig. 3. Spectra of carbon fiber fabric and g-C<sub>3</sub>N<sub>4</sub>: (a) XRD; (b) FTIR; (c) XPS C1s; (d) TGA.

**Table 1**  
Surface element composition of carbon fiber fabric and g-C<sub>3</sub>N<sub>4</sub>.

Sample	Element content (%)				
	C	O	N	O/C	N/C
CFs	88.59	10.23	1.18	11.55	1.33
g-C <sub>3</sub> N <sub>4</sub>	44.18	7.19	48.63	16.27	110.07
CFs-C <sub>3</sub> N <sub>4</sub>	50.68	5.79	43.53	11.42	85.89

absorption peaks centered at 806 and 2178 cm<sup>-1</sup> were assigned to the breathing mode of s-triazine and C≡N triple bonds, respectively [58]. The characteristic absorption peaks in the range of 3500–3000 cm<sup>-1</sup> were attributed to the stretching mode of O–H bond and the vibration mode of N–H bond [59]. Interestingly, the spectrum of CFs-C<sub>3</sub>N<sub>4</sub> displays a similar absorption peak in the wavenumber range of 4000–400 cm<sup>-1</sup> due to the formation of g-C<sub>3</sub>N<sub>4</sub> on the carbon fiber surface. The increased surface functional groups on CFs-C<sub>3</sub>N<sub>4</sub> surface can effectively improve the interfacial compatibility and strengthen the interfacial bonding between matrix and reinforcement.

In addition, the surface element composition of carbon fiber fabric was investigated by XPS. The detailed information on surface element composition of samples was summarized in Table 1. The surfaces of all samples are composed of carbon, oxygen and nitrogen element. Carbon is the main element on CF surface (88.59%), and the content of carbon and nitrogen on g-C<sub>3</sub>N<sub>4</sub> surface is almost equal (44.18 and 48.63%, respectively). The formation of g-C<sub>3</sub>N<sub>4</sub> results in a vast increase in the nitrogen content on the surface of CF fabric (increased by 35.89 times). Moreover, the XPS C1s spectra were also supplied for further understanding the chemical element state on CF surface. As shown in Fig. 3(c), the characteristic peak at about 284.6 eV for CFs sample reflects the sp<sup>2</sup> hybridized C atom in the C–C group [52]. Meanwhile, the XPS C1s spectra of g-C<sub>3</sub>N<sub>4</sub> reveal two different characteristic peaks located at 284.6 and 288.1 eV, respectively. The characteristic peak centered at 288.1 eV is attributed to the sp<sup>2</sup>-bonded C in N=C (–N)<sub>2</sub> group [60]. More significantly, the characteristic peak of sp<sup>2</sup>-bond C in N=C (–N)<sub>2</sub> group appeared in the C1s spectra of CFs-C<sub>3</sub>N<sub>4</sub> after the formation of g-C<sub>3</sub>N<sub>4</sub> on CF fabric surface. Moreover, the existence of g-C<sub>3</sub>N<sub>4</sub> was also proved by TGA test (Fig. 3(d)). Owing to the thermal degradation of surface materials in the temperature range of 50–800 °C, a weight loss of 3.19% was observed in the TGA curve of CFs. The TGA analysis of CFs-C<sub>3</sub>N<sub>4</sub> embodies a higher weight loss of 6.65% due to the decomposition of g-C<sub>3</sub>N<sub>4</sub>. Therefore, a conclusion can be drawn from the above XPS and TGA analysis that g-C<sub>3</sub>N<sub>4</sub> was successfully synthesized on the CF fabric surface.

### 3.3. Interfacial properties

The interfacial compatibility (or wettability) between reinforcement and matrix is generally affected by the surface energy of reinforcement. High surface energy equals to a better interfacial compatibility and stronger interfacial adhesion. Here, the surface energy of carbon fiber samples was measured by a Cahn dynamic angle analysis system. Table 2 provides a summary and comparison of contact angle ( $\theta$ ), surface energy ( $\gamma$ ) and its sub-component (polar component:  $\gamma^p$  and dispersion component:  $\gamma^d$ ). The de-sizing treatment of CFs causes a

**Table 2**  
Contact angle, surface energy and roughness of carbon fibers.

Sample	Contact angle (°)		Surface energy (mJ/m)		
	Deionized water	Diiodomethane	$\gamma^d$	$\gamma^p$	$\gamma$
As-received CFs	67.35	58.14	29.65	12.25	41.90
CFs	77.25	61.47	27.73	7.72	35.45
CFs-C <sub>3</sub> N <sub>4</sub>	48.81	36.13	41.50	17.99	59.49

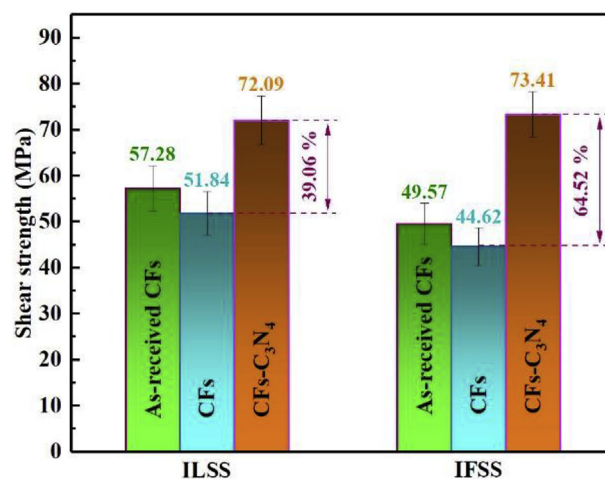


Fig. 4. IFSS and ILSS of carbon fiber and its composites laminates.

certain degree of reduction of surface energy from 41.90 to 35.45 mJ/m. Surprisingly, CFs-C<sub>3</sub>N<sub>4</sub> exhibits a smaller contact angles and higher surface energy compared to that of CFs. The in-situ synthesis of g-C<sub>3</sub>N<sub>4</sub> leads to the contact angle of CFs with deionized water (polar solvent) decreasing from 77.25° to 48.81°, the contact angle of CFs with diiodomethane (non-polar solvent) decreased from 61.47° to 36.13°. This change directly causes the surface energy of CFs-C<sub>3</sub>N<sub>4</sub> rise from 35.45 to 59.49 mJ/m with an increase rate of 67.81%. The growth of  $\gamma^p$  (from 7.72 to 17.99 mJ/m) and  $\gamma^d$  (from 27.73 to 41.50 mJ/m) was attributed to the substantially increased surface roughness and functional groups created by g-C<sub>3</sub>N<sub>4</sub>. It is totally expected that the hierarchical structure of CFs-C<sub>3</sub>N<sub>4</sub> with a high surface energy can more effectively ameliorate the interfacial compatibility between CFs and epoxy resin, thus significantly improves the interfacial properties of composites.

The good interfacial properties can endow composites with excellent mechanical properties. Here, ILSS as an important index of mechanical properties of CF composite laminates was assessed by three-point bending test. As illustrated in Fig. 4, compared with as-received CFs, the ILSS of CFs decreases in a certain extent (from 57.28 to 51.84 MPa) after de-sizing treatment. The introduction of g-C<sub>3</sub>N<sub>4</sub> led to the ILSS of composite laminates made of CFs-C<sub>3</sub>N<sub>4</sub> increase to 72.09 MPa with a rise rate of 39.06%. Besides, the IFSS as the frequent reference standard for interfacial adhesion between CFs and epoxy resin was also studied by micro-droplet debonding experiment. The IFSS exhibits the similar change trend with that of ILSS. The IFSS of CF composite laminates decreases firstly from 49.57 to 44.62 MPa due to de-sizing treatment, then increases to 73.41 MPa with a growth rate of 64.52% after the in-situ treatment. The significantly improved interfacial properties of composite laminates can be attributed to the massive increased mechanical interlocking and chemical bonding in the interphase formed during the surface modification [61]. The introduced g-C<sub>3</sub>N<sub>4</sub> creates a better interfacial compatibility, bigger contact area and stronger mechanical interlocking for the interfacial bonding. Moreover, the open-ring reaction between amino groups on the g-C<sub>3</sub>N<sub>4</sub> surface and epoxy groups of matrix provides a stronger chemical bonding for interfacial adhesion of composites.

### 3.4. Mechanical properties

The mechanical properties of CFs and its composite laminates were investigated using tensile and bending test. As shown in Fig. 5, the tensile strength (TS) of single carbon fiber for as-received CFs, CFs and CFs-C<sub>3</sub>N<sub>4</sub> is 3.49, 3.47 and 3.56 GPa, respectively. The treatment process has a negligible impact on the mechanical properties of CFs. Besides, a comparison of mechanical properties for carbon fiber composite laminates was supplied in Fig. 5(b). The introduction of g-C<sub>3</sub>N<sub>4</sub> on

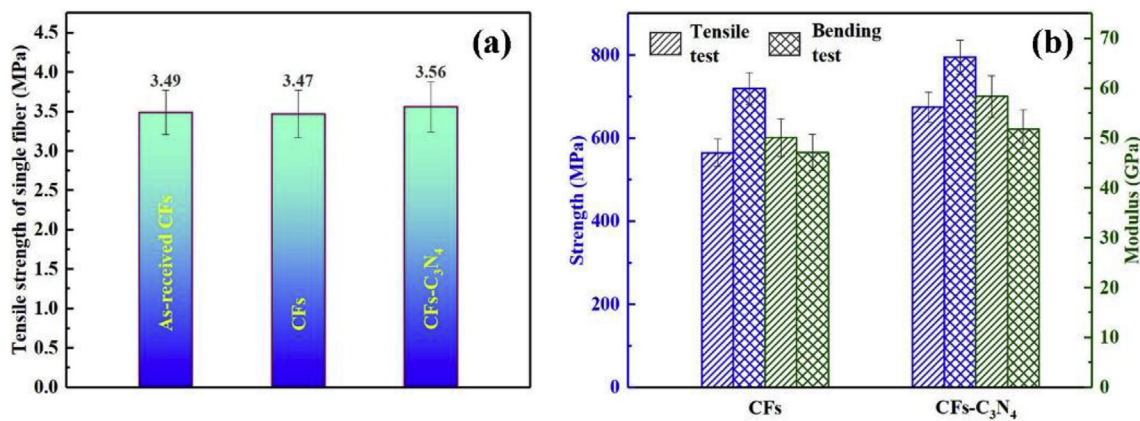


Fig. 5. Tensile test and bending test for carbon fibers fabric composite laminates.

carbon fiber fabric surface significantly enhanced the mechanical properties of carbon fiber composite laminates. The TS and tensile modulus of CF composite laminates are increased from 564.91 MPa to 675.31 MPa (increased by 19.54%) and from 50.12 GPa to 58.37 GPa (increased by 16.46%), respectively. Meanwhile, the bending strength and bending modulus have also been improved from 719.51 to 795.13 MPa (rise by 10.51%) and from 47.14 to 51.83 GPa (rise by 9.95%). The significantly enhanced mechanical properties of carbon fiber composite laminates can be attributed to the synergistic effects of increased wettability, mechanical interlocking and chemical bonding between CFs and epoxy resin result from the introduction of g-C<sub>3</sub>N<sub>4</sub> [62].

Here, impact properties as an important index of mechanical properties of carbon fiber composite laminates were also explored. As illustrated in Fig. 6, the initial, propagative and total absorbed energy of CFs composite laminate was 0.21, 0.93 and 1.14 J, respectively. The impact properties of CFs-C<sub>3</sub>N<sub>4</sub> composite laminates were greatly enhanced duo to the in-situ synthesis of g-C<sub>3</sub>N<sub>4</sub>. The initial, propagative and total absorbed energy was increased to 0.54, 1.24 and 1.78 J, respectively. Generally, the interphase of carbon fiber composite laminates as a protective layer can absorb the fracture energy through triggering cracks and sub-cracks, thus holds back the direct contact between crack tips and CFs [63]. Obviously, the complex structure of interphase containing g-C<sub>3</sub>N<sub>4</sub> can cause more cracks and sub-cracks, dissipate more destructive energy, thus effectively improves the impact failure resistance. Meanwhile, the failure surface morphology of carbon fiber composite laminates was further explored for impact properties

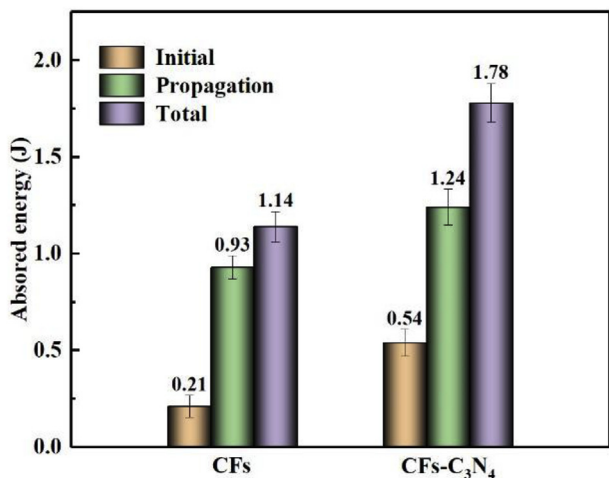


Fig. 6. Impact properties of carbon fibers composite laminates reinforced with CFs and CFs-C<sub>3</sub>N<sub>4</sub>, respectively.

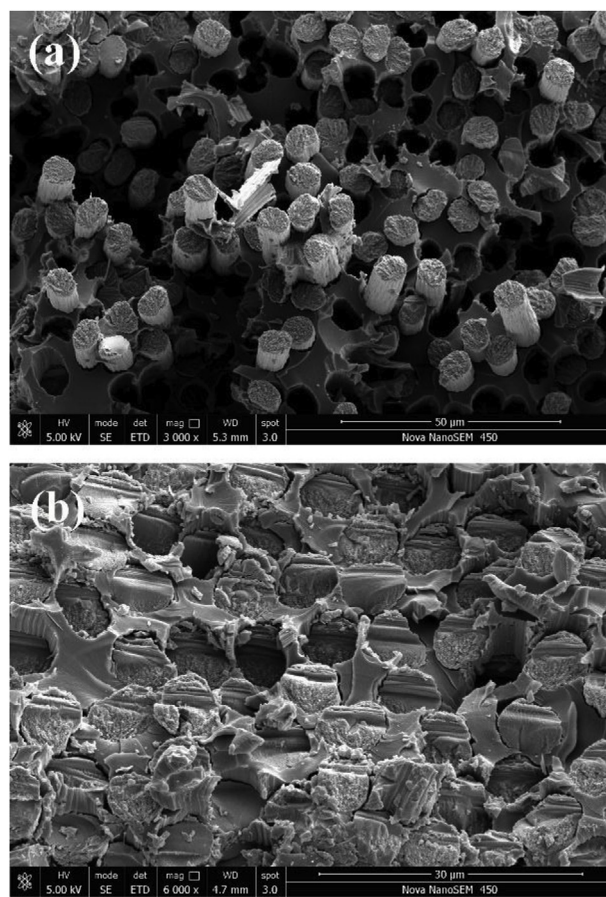


Fig. 7. Failure surface morphology for carbon fibers composite laminates: (a) CFs and (b) CFs-C<sub>3</sub>N<sub>4</sub>.

(Fig. 7). Interfacial de-bonding, cracks and pull-out of fiber are found on the fracture surface of CF composite laminates due to poor interfacial adhesion. The flat fracture morphology of CFs-C<sub>3</sub>N<sub>4</sub> indicates a stronger interfacial adhesion supplied by better wettability, increased mechanical interlocking and enhanced chemical bonding exists in the interphase of composite laminates.

Base on the above analysis, an impact test model was depicted in Fig. 8 for revealing the mechanism of interphase in the composite laminates. For CF composite laminates, inferior interfacial adhesion created by simple structure of the thin interphase easily triggers a low effective load transfer between reinforcement and matrix, and a faster propagation of cracks along the interfacial area under a low applied

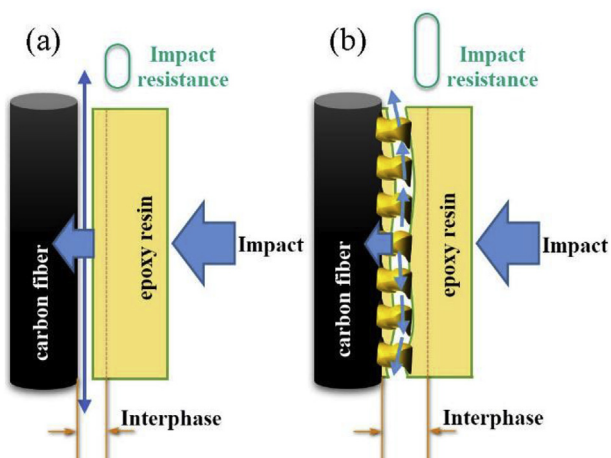


Fig. 8. Schematic impact test of carbon fibers composite laminates: (a) CFs and (b) CFs-C<sub>3</sub>N<sub>4</sub>.

load. In the case of CFs-C<sub>3</sub>N<sub>4</sub> composite laminates, the powerful and complicated structure of the thick interphase provides stronger interfacial chemical bonding and mechanical interlocking force, thus effectively ameliorate the interfacial load transfer, promptly release the stress concentration and sharply dissipate the destructive energy [64]. Moreover, the complicated cracks propagation and improved deformation ability of matrix can promote the formation of micro-cracks and change of failure mode [65]. This complex propagation process of cracks can markedly enhance the impact properties through consuming more energy.

Furthermore, the dynamic mechanical properties were also investigated. Fig. 9 indicates the variation of storage modulus ( $E'$ ) and loss factor ( $\tan \delta$ ) with temperature for carbon fiber composite laminates. Clearly, CFs-C<sub>3</sub>N<sub>4</sub> composite laminates have a higher  $E'$  than that of CFs composite laminates. The introduction of g-C<sub>3</sub>N<sub>4</sub> causes  $E'$  increase from 37.17 to 43.83 GPa below glass transition temperature ( $T_g$ ) and from 7.65 to 16.40 GPa above  $T_g$ , which embodies a superior stability of stiffness at near  $T_g$ . This increased  $E'$  was ascribed to the enlarged interphase region and reduced mobility of polymer chains in the interphase [66]. Hence, the complex interphase constructed by g-C<sub>3</sub>N<sub>4</sub> can act as an auxiliary reinforcement for mechanical stiffening of carbon fiber composite laminates. Interestingly, similar phenomena and conclusions were obtained in the CNT/CF composite system [67]. Meanwhile, the  $T_g$  of carbon fiber composite laminates was raised from 109.71 to 121.34 °C after the surface treatment. The enhancement of  $T_g$  can be attributed to the strengthened restricted-mobility of polymer chains in the interphase [68]. The high specific surface area and abundant surface functional groups of g-C<sub>3</sub>N<sub>4</sub> significantly enhance the wettability and interaction between CFs and epoxy resin and increase the crosslinking degree in the interphase. Furthermore, the g-C<sub>3</sub>N<sub>4</sub> can also restrain the viscous flow of polymer chains, thereby realize the energy loss minimization of viscous deformation.

### 3.5. Hydrothermal aging resistance

The performance and service life of CF composite laminates are often affected by the surrounding environment. For example, high humidity environments can hydrolyze and damage the interphase structure of CF composites, thus degrade the whole performance and life of composites. Here, hydrothermal aging resistance as an important embodiment of performance stability of CF composite laminates was evaluated. As illustrated in Fig. 10, a serious degradation in ILSS occurred after hydrothermal aging treatment. The IFSS of CFs and CFs-C<sub>3</sub>N<sub>4</sub> was reduced by 31.29% (from 51.84 to 35.62) and 18.89% (from 72.09 to 58.47), respectively. The interfacial adhesion of CFs composite

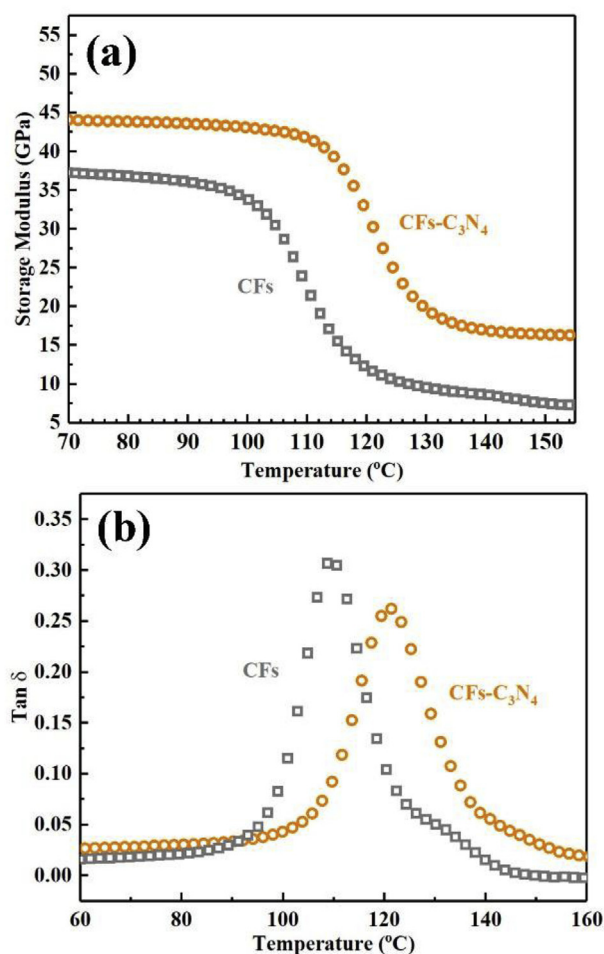


Fig. 9. Dynamic mechanical analysis of carbon fibers composite laminates: (a) storage modulus and (b)  $\tan \delta$ .

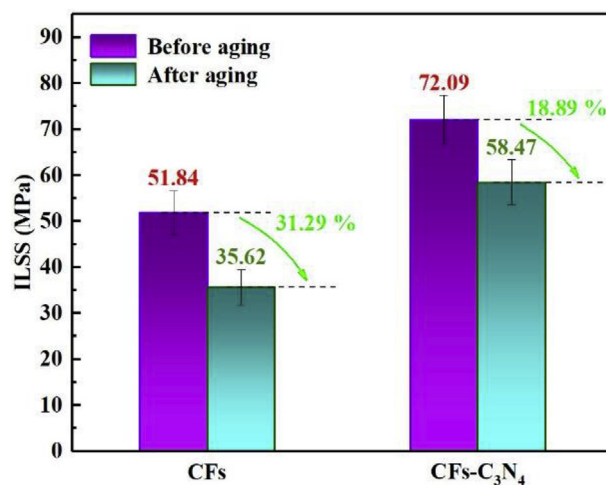


Fig. 10. Change of ILSS before and after hydrothermal aging for 48 h.

laminates is so poor that water molecules can easily penetrate into the interphase, thus severely destroy the interfacial structure and substantially degrade the interfacial properties [69,70]. The markedly improved hydrothermal aging resistance of CFs-C<sub>3</sub>N<sub>4</sub> composite laminates was ascribed to the strong and powerful interfacial adhesion created by the increased mechanical interlocking and chemical bonding between CFs and epoxy resin [71–74]. In addition, the damage of C–N

bond created by opening-ring reaction between amino groups on g-C<sub>3</sub>N<sub>4</sub> surface and epoxy groups of matrix can consume more destructive energy, thus enhances the structural stability of interphase [75–99].

#### 4. Conclusions

Here, g-C<sub>3</sub>N<sub>4</sub> was used to structure the interphase of carbon fiber composite laminates for interfacial strengthening via an in-situ synthesis. The existence of g-C<sub>3</sub>N<sub>4</sub> greatly increased surface roughness, surface functional groups and surface free energy of carbon fibers, thus significantly enhance the interfacial properties of carbon fibers composite. Surface free energy was increased from 35.45 to 59.49 mJ/m. The ILSS and IFSS of carbon fiber composite laminates were enhanced from 51.84 to 72.09 MPa and 44.62–73.41 MPa, respectively. The significantly improved interfacial properties guarantee excellent mechanical properties of carbon fiber composite laminate. Tensile strength and bending strength of composite laminates were increased from 564.91 to 675.31 MPa and 719.51–795.13 MPa, respectively. Total absorbed energy of impact test was increased from 1.14 to 1.78 J. Furthermore, dynamic mechanical properties and hydrothermal aging resistance were also improved significantly. The  $T_g$  was increased from 109.71 to 121.34 °C. The  $E'$  of composite laminates was increased from 37.17 to 48.83 GPa below  $T_g$  and from 7.65 to 16.40 GPa above  $T_g$ , respectively. The hydrothermal aging test confirms that the ILSS of carbon fiber composite laminates was decreased from 31.29 to 18.89%. The markedly enhanced interfacial properties and mechanical properties of CFS-C<sub>3</sub>N<sub>4</sub> composite laminates can be attributed to the increased mechanical interlocking, enhanced chemical bonding and ameliorated wettability due to the existence of g-C<sub>3</sub>N<sub>4</sub> on carbon fibers surface. With the inherent lightweight as compared with metals and ceramics, the surface modification mentioned in this study not only can be used for the interfacial strengthening of composite laminates, but also can be applied in other fields, such as environmental governance, energy development and ultra-capacitor [78–82].

#### Declaration of interest

The authors declare no conflict of interest regarding the publication of this paper.

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